

Dichloro[tris(methylpyridyl)amine]cobalt(III)
perchlorate hemihydrate

Timothy W. Failes

Centre for Heavy Metals Research, School of
Chemistry, University of Sydney, NSW 2006,
AustraliaCorrespondence e-mail:
t.failes@chem.usyd.edu.au

Key indicators

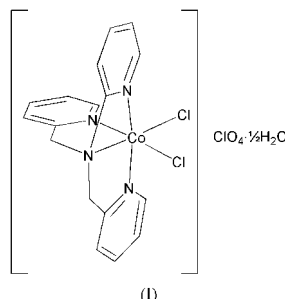
Single-crystal X-ray study
 $T = 150$ K
Mean $\sigma(\text{C}-\text{C}) = 0.002$ Å
Disorder in solvent or counterion
 R factor = 0.025
 wR factor = 0.074
Data-to-parameter ratio = 26.9For details of how these key indicators were
automatically derived from the article, see
<http://journals.iucr.org/e>.

The tris(methylpyridyl)amine (tpa, $\text{C}_{18}\text{H}_{18}\text{N}_4$) ligand in the title complex, $[\text{Co}(\text{tpa})\text{Cl}_2]\text{ClO}_4 \cdot 0.5\text{H}_2\text{O}$, is bound to the Co^{III} centre in a tripodal fashion through the four N-donor atoms, allowing the two chloro ligands to occupy the remaining *cis* sites in the octahedral coordination sphere. Half a solvent water molecule is disordered over two sites. The metal–ligand bond lengths are typical for Co^{III} complexes.

Received 30 May 2003
Accepted 30 June 2003
Online 24 July 2003

Comment

The title complex, (I) (Fig. 1 and Table 1), is the first crystal structure of a Co^{III} complex containing the tris(methylpyridyl)amine (tpa) ligand. The complex is useful as a synthetic precursor to a wide variety of Co^{III} –tpa complexes, particularly those with bidentate chelating ligands. Furthermore, the structural parameters may be of use for molecular mechanics force fields, given the interest in studying Co^{III} –N systems with such methods.



Experimental

The complex was prepared according to a reported method (Hall *et al.*, 2003). Crystals of the complex were grown by slow cooling of an ethanol–methanol (1:1) solution of the complex.

Crystal data

$[\text{CoCl}_2(\text{C}_{18}\text{H}_{18}\text{N}_4)]\text{ClO}_4 \cdot 0.5\text{H}_2\text{O}$
 $M_r = 527.64$
Monoclinic, $P2_1/c$
 $a = 10.2642$ (14) Å
 $b = 15.166$ (2) Å
 $c = 14.2695$ (19) Å
 $\beta = 109.001$ (3)°
 $V = 2100.2$ (5) Å³
 $Z = 4$

$D_x = 1.669$ Mg m⁻³
Mo $K\alpha$ radiation
Cell parameters from 172
reflections
 $\theta = 2.0$ – 33.0 °
 $\mu = 1.24$ mm⁻¹
 $T = 150$ (2) K
Prism, purple
 $0.50 \times 0.30 \times 0.10$ mm

Data collection

Bruker SMART CCD
diffractometer
 φ and ω scans
Absorption correction: multi-scan
(SADABS; Sheldrick, 1996)
 $T_{\text{min}} = 0.577$, $T_{\text{max}} = 0.886$
31 017 measured reflections

7781 independent reflections
6722 reflections with $I > 2\sigma(I)$
 $R_{\text{int}} = 0.025$
 $\theta_{\text{max}} = 33.0$ °
 $h = -15 \rightarrow 15$
 $k = -23 \rightarrow 23$
 $l = -21 \rightarrow 21$

Refinement

Refinement on F^2
 $R[F^2 > 2\sigma(F^2)] = 0.025$
 $wR(F^2) = 0.074$
 $S = 1.03$
 7781 reflections
 289 parameters
 H-atom parameters constrained

$$w = 1/[\sigma^2(F_o^2) + (0.0426P)^2 + 0.434P]$$

where $P = (F_o^2 + 2F_c^2)/3$
 $(\Delta/\sigma)_{\max} < 0.001$
 $\Delta\rho_{\max} = 0.51 \text{ e } \text{\AA}^{-3}$
 $\Delta\rho_{\min} = -0.41 \text{ e } \text{\AA}^{-3}$

Table 1

Selected geometric parameters (Å).

Co1—N1	1.9450 (10)	Co1—N4	1.9624 (9)
Co1—N2	1.9553 (9)	Co1—Cl1	2.2374 (4)
Co1—N3	1.9357 (9)	Co1—Cl2	2.2630 (4)

The half-water molecule is disordered over two sites with occupancy factors fixed at 0.25 each. H atoms were placed in calculated positions using a riding model, with U_{eq} set at $1.2U_{\text{eq}}$ of the parent atom. H atoms of the disordered water molecule were not included.

Data collection: *SMART* (Bruker, 1995); cell refinement: *SMART*; data reduction: *SAINT-Plus* (Bruker, 1995); program(s) used to solve structure: *SHELXS97* (Sheldrick, 1997); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997); molecular graphics: *ORTEP-3 for Windows* (Farrugia, 1997); software used to prepare material for publication: *WinGX* publication routines (Farrugia, 1999).

References

- Bruker (1995). *SMART* and *SAINT-Plus*. Bruker AXS Inc., Madison, Wisconsin, USA.
 Farrugia, L. J. (1997). *J. Appl. Cryst.* **30**, 565.
 Farrugia, L. J. (1999). *J. Appl. Cryst.* **32**, 837–838.

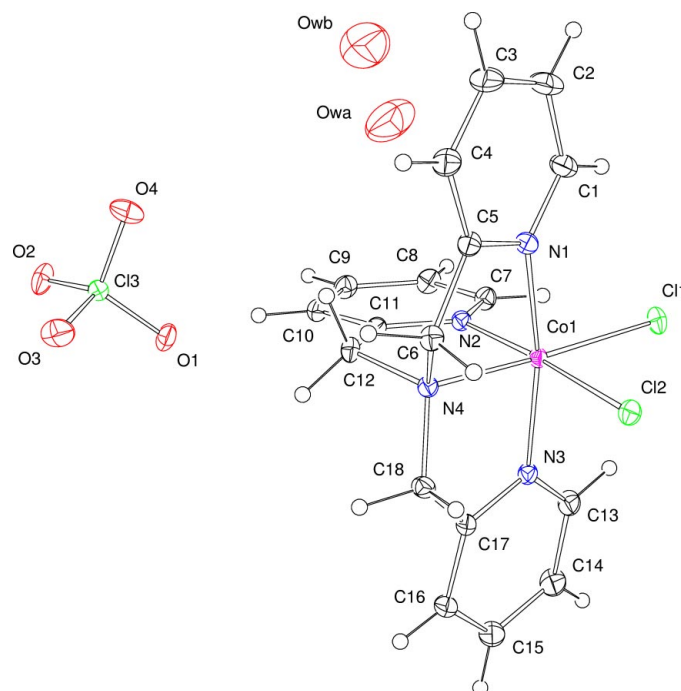


Figure 1

Displacement ellipsoid plot of (I), showing the crystallographic numbering scheme. Displacement ellipsoids are drawn at the 30% probability level.

- Hall, M. D., Underwood, C. K., Failes, T. W., Zhang, M., Hambley, T. W. (2003). *J. Inorg. Biochem.* Submitted.
 Sheldrick, G. M. (1996). *SADABS*. University of Göttingen, Germany.
 Sheldrick, G. M. (1997). *SHELXS97* and *SHELXL97*. Release 97-2. University of Göttingen, Germany.